- 1 Endosulfan, pentachlorobenzene and short-chain chlorinated paraffins in background soils
- 2 from Western Europe
- 3 Anne Karine Halse^{1,2*}, Martin Schlabach¹, Jasmin K. Schuster³, Kevin C. Jones³, Eiliv Steinnes⁴,
- 4 Knut Breivik^{1,5}
- ¹ Norwegian Institute for Air Research, Box 100, NO-2027 Kjeller, Norway
- 6 ² Norwegian University of Life Sciences, Department of Chemistry, Biotechnology and Food
- 7 Science, P.O. Box 5003, NO-1432 Ås, Norway
- 8 ³ Lancaster Environment Centre, Lancaster University, Lancaster, LA1 4YQ, UK
- 9 ⁴ Norwegian University of Science and Technology, Department of Chemistry, NO-7491
- 10 Trondheim, Norway
- ⁵ University of Oslo, Department of Chemistry, Box 1033, NO-0315 Oslo, Norway
- *Corresponding author. Norwegian Institute for Air Research, NO-2027 Kjeller, Norway.
- 13 Tel.: +47 63 89 80 00; fax: +47 63 89 80 50. E-mail: <u>akh@nilu.no</u> (A.K. Halse)

14 **Abstract**

Soils are major reservoirs for many persistent organic pollutants (POPs). In this study, "newly" 15 16 regulated POPs i.e. Σ endosulfans (α -endosulfan, β -endosulfan, endosulfan sulfate), pentachlorobenzene (PeCB), and short-chain chlorinated paraffins (SCCPs) were determined 17 in background samples from woodland (WL) and grassland (GL) surface soil, collected along 18 19 an existing latitudinal UK-Norway transect. Statistical analysis, complemented with plots 20 showing the predicted equilibrium distribution and mobility potential, was then explored to 21 discuss factors controlling their spatial distribution. SCCPs were detected with the highest 22 average concentrations (2435±72100 ng/g soil organic matter (SOM)), followed by Σendosulfans ($\frac{32\pm43}{2}$ ng/g SOM) and PeCB ($\frac{40.6\pm0.54}{2}$ ng/g SOM). PeCB and Σendosulfans 23 share many similarities in their distribution in these background soils as well as with several 24 25 legacy POPs. A steep decline in concentrations of SCCPs with increasing latitude indicates that their occurrence is dictated by proximity to source regions, while concentrations of 26 27 Σ endosulfans peaked in regions experiencing elevated precipitation rates.

- 28 Capsule
- 29 Processes affecting the occurrence and latitudinal distribution of some "newly" regulated
- 30 POPs in background soils.

Keywords

- 32 Short-chain chlorinated paraffins, Endosulfan, Pentachlorobenzene,
- 33 Background soils, Soil organic matter

1 Introduction

Soils have a large affinity and capacity for sorbing and storing persistent organic pollutants (POPs) (Meijer et al., 2003). Yet, the storage capacities of soils are expected and observed to vary significantly across the globe, reflecting differences in e.g. proximity to global source regions of POPs, soil organic matter content, and climatic conditions (Meijer et al., 2003). Soils from boreal regions are of particular interest, due to the low temperatures, high organic matter content, and slow decomposition associated with this type of environment (Moeckel et al., 2008). Hence, soils from boreal regions are expected to have elevated storage capacities for POPs (Dalla Valle et al., 2005).

Soils located remote from potential sources such as e.g. industrial or populated regions and agricultural areas are defined as background soils (Meijer et al., 2003). The POP contamination of background soils is mainly due to atmospheric deposition. Background soils may therefore be utilized to study potential pollution gradients as mediated by atmospheric transport, e.g. from potential source regions towards remote areas. A latitudinal transect of background soil sites, extending from southern parts of the UK towards northern Norway (UK-Norway transect), was established in 1998 (Meijer et al., 2002). Past studies from the UK-Norway transect have reported data on the occurrence and distribution of legacy POPs such as polychlorinated biphenyls (PCBs) (Meijer et al., 2002; Schuster et al., 2011), polybrominated diphenyl ethers (PBDEs) (Hassanin et al., 2004), polycyclic aromatic hydrocarbons (PAHs) (Nam et al., 2008b), polychlorinated dibenzodioxins and dibenzofurans (PCDDs/Fs) (Hassanin et al., 2005) and selected organochlorine pesticides (Schuster et al., 2011). While the UK-Norway transect has been considered a suitable "model environment" for empirical evaluation of longrange atmospheric transport (LRAT) behaviour of legacy POPs, studies of "new" (i.e. recently regulated) POPs have still been lacking. The overall goal of the study is to obtain a quantitative

and mechanistic understanding of the occurrence, fate, and distribution in background soils from North-western Europe of the following three substances/substance groups, collectively referred to as "new" POPs in this study.

- 62 (i) Endosulfan is an organochlorine pesticide, which is banned in many parts of the world.
 63 Technical mixtures of endosulfan contain two isomers, α- and β-endosulfan (α/β) in a
 64 7/3 or 2/1 ratio, depending on the origin of the technical mixture (Weber et al., 2010).
 65 Technical endosulfan was listed under the Stockholm Convention on POPs (Annex A) in
 66 2011 (UNEP, 2011).
- 67 (ii) Pentachlorobenzene (PeCB) has been used in production of the pesticide quintozene
 68 (pentachloronitrobenzene) and in a variety of chlorobenzene mixtures in order to
 69 reduce the viscosity of dielectric fluids. Current sources of PeCB to the environment
 70 include various combustion and industrial processes, as well as degradation of other
 71 chemicals (Bailey et al., 2009). PeCB was listed in Annex A and C under the Stockholm
 72 Convention in 2009 (UNEP, 2009), in addition to being adopted under the 1998 Aarhus
 73 protocol in 2009 (UNECE, 2010).
 - (iii) Short-chain chlorinated paraffins (SCCPs) are a complex mixture of polychlorinated *n*-alkanes. The alkane chain length can reach up to 13 carbons with chlorine content ranging between 30-70% (Houde et al., 2008). SCCPs have the general chemical formula C_xCl_yH_{2x+2-y} where x is the number of carbon atoms (x=10-13) and y is the number of chlorine atoms (y=1-13) (Muir et al., 2000). SCCPs have been utilized in e.g. metal-cutting fluids, sealants, paints, lubricants, flame-retardants, and as softeners in different materials (UNECE ad hoc expert group on POPs, 2003; van der Gon et al., 2007). They have a potential to be released into the environment through production, storage, leaching, and/or volatilization (Muir et al., 2000). SCCPs have been adopted under the 1998 Aarhus protocol (UNECE, 2010) and are being evaluated by the Persistent Organic Pollutants Review Committee (POPRC) under the Stockholm Convention on POPs (UNEP, 2012).
 - There are few studies which have reported levels and distribution of these new POPs in background soils. However, endosulfans have previously been studied in forest soils in Canada and UK (Wong et al., 2009) and Tajikistan (Zhao et al., 2013), PeCB has been measured in

Estonian (Roots et al., 2010) and Antarctic soils (Borghini et al., 2005) while SCCPs have been reported in background soils from China (Wang et al., 2013).

2 Materials and methods

2.1 Sampling

Background surface soil samples were collected in 2008 (Schuster et al., 2011), along an existing transect of 39 sites (Meijer et al., 2002). This transect includes a number of United Kingdom (UK) sites (n=16) and Norwegian sites (n=23), covering remote/rural areas from 50.58-70.47° N and -6.20-27.95° E. Site descriptions are provided in Table A1 while a map is provided in Figure A1 in the Supplementary Data (SD). The soil sampling strategy together with information on soil characteristics has been presented earlier (Schuster et al., 2011), and only a brief summary is given here. The soil samples were collected to represent two types of remote soils, i.e. grassland soils (GL, n=25) and woodland (coniferous/deciduous) soils (WL, n=34). 59 surface soil samples in total were sampled and analysed. The samples were collected with a stainless steel hand-held corer and the targeted sampling depth was the upper layer (0-5 cm). Ten cores were taken to cover an area over several square meters and pooled to obtain a composite sample from each site (Meijer et al., 2002). The lumped samples were placed in zip-locked bags and stored in a freezer at ~-18 °C until further sample preparation.

2.2 Sample preparation, clean-up, and analysis

In the following section, a short overview of the sample treatment is given. More detailed information of the sample preparation, including drying, sieving, extraction, clean-up and analysis of the soil samples, is available in the SD (Sect A2.1-A2.3).

Wet soil samples were placed on aluminium foil and dried in an oven at low temperature (~27 °C), to minimize the potential for volatilization of target compounds. The time required to obtain complete dryness was approximately two weeks. Diatomaceous earth (DE) was prepared along with the soil samples, in order to address possible laboratory contamination from drying and clean-up procedures. Dried soil samples were afterwards sieved with a precleaned sieve and stored cold in clean, brown glass bottles, covered with aluminium foil and a polypropylene (PP)-lid until extraction.

Extraction of the soil samples was performed by use of a Dionex Accelerated Solvent Extractor (ASE-200, 1995) unit (California, USA). An ASE cell was packed with soil samples that had been pre-mixed with DE for better extraction efficiency (see Figure A2). In addition, a mixture of internal standard was added. The extraction was carried out with acetone/*n*-hexane 1/1 (w/w). The content of soil organic matter (SOM) was determined by loss on ignition (LOI) at 550 °C (A2.6).

The soil extracts were divided into two aliquots prior to the clean-up procedures (A2.2.2), i.e. an endosulfan fraction (α -endosulfan, β -endosulfan and endosulfan sulfate), referred collectively to as Σ -endosulfans, and an acid stable fraction (PeCB, SCCPs). The aliquot for the determination of Σ -endosulfans was cleaned using a silica column only, while the aliquot for the determination of the acid stable compounds was treated with sulphuric acid followed by a fractionation with a silica column (A2.2.2). Subsequently, both aliquots were reduced to ~50 μ l by nitrogen and recovery standards were added (see A2.3). There were some problems with sedimentation in the endosulfan extracts, and it was necessary to centrifuge the samples (see A2.2.2) prior to instrumental analysis. Σ -Endosulfans and SCCPs were determined with high resolution gas chromatography coupled to mass spectrometry in an electron capture negative ion mode (GC/HRMS(ECNI)), while PeCB was determined using gas chromatography coupled to high resolution mass spectrometry in an electron impact (EI) mode (GC/HRMS(EI)) (see Table A2). More information concerning the general method validation (QA/QC) is given in SD (A2.4 and A2.5).

2.3 Statistical analysis

Statistical analysis was performed with Microsoft Office Excel 2007 (Albuquerque, New Mexico, USA). Pearson correlation coefficients (r) were calculated to evaluate any linear relationships between the included variables, while the significance (p) was estimated by use of a Student's t-distribution (see A2.8). A correlation with a p < 0.05 was defined as statistically significant. For statistical summaries of overall results, samples with concentrations below method detection limit (MDL) (see A2.5) were assigned a value of $\frac{1}{2}$ MDL. Prior to the correlation analyses, data below MDL and outliers were removed. The dataset was reviewed for possible outliers using a z-score test, and a z-score value ≥ 3 was determined as an outlier (A2.8.1). Additional parameters explored in the statistical analysis were bulk density (BD),

black carbon (BC), SOM content, and concentrations of HCB, PBDEs and PCBs which were all adapted from Schuster et al. (2011), with the exception of SOM. All data, with the exception of values for latitude and land use, were log transformed prior to the correlation analysis.

2.4 Mapping phase distribution and fate in soils

149

150

151

152

153

154

155

156

157

158

159

160

161

162

163

164

165

166

167

168

169

170

171

172

173

174

175

176

177

178

The distribution of organic compounds within environmental compartments is largely determined by their environmental partitioning behaviour which, in turn, is dictated by their physical-chemical properties (Gouin et al., 2000). Following Wong and Wania (2011), the phase distribution and mobility potential of individual substances in soils can therefore be plotted in chemical partitioning space maps (Figure 1). This simple graphical technique facilitates mechanistic insights into the likely partitioning and fate in soils and is used to complement interpretation of measurement data and results obtained based on the statistical analysis. Physical-chemical properties used to represent the various substances were adopted from the literature. Data for PCBs and PBDEs (Li et al., 2003; Wania and Dugani, 2003), HCB, PeCB, α -, β -endosulfan (Shen et al., 2005), endosulfan sulfate (U.S. EPA, 2011) and SCCPs (European Chemicals Bureau, 2008; Gawor and Wania, 2013)). The SCCPs are represented by a 38 formula groups, each representing the average properties of all isomers within each group (Krogseth et al., 2013), as well as the average property derived for SCCPs in the European Union (EU) Risk Assessment Report (RAR) (European Chemicals Bureau, 2008). For details, see SD (sect. A2.7). However, we caution that uncertainties remain with regards to the physical-chemical properties for SCCPs and refer to Gluege et al., (2013) for a detailed analysis and discussion.

3 Results and discussion

3.1 Overall results

Table 1 presents a summary of overall results for Σ endosulfans and its individual constituents as well as for PeCB and SCCPs, expressed as ng/g SOM. Table A4 shows the same results, but expressed on a ng/g dw basis, while Table A5 presents concentrations for individual compounds and sampling sites, expressed as ng/g dw and ng/g SOM. A few results were considered invalid and excluded from further analysis, either because of matrix related disturbances (Σ endosulfans; n=2, SCCPs and PeCB; n=1) or a low recovery of less than 10% (PeCB; n=5) (see also A2.4). Due to the complexity of SCCP mixture, a complete separation of

the individual congeners was not achieved (Sverko et al., 2012; Tomy et al., 1997), consequently introducing higher uncertainty into the quantified results of this compound group. The results for SCCPs should therefore be interpreted with caution. Concentrations were often below MDL for several of the studied compounds (see A2.5). Following removal of invalid data and data below MDL, one or more endosulfan constituents were detected in 56 out of the 57 samples considered valid (Table A5).

Within the group of Σ endosulfans, endosulfan sulfate was most frequently detected (n=56), followed by β -endosulfan (n=17) and α -endosulfan (n=13) (Table 1). PeCB was detected in 52 samples, while SCCPs were detected in 23 samples. The average concentration of SCCPs analysed (n=58) was by far the highest among the "new" POPs with 2435 ng/g SOM (SD±10072 ng/g SOM), ranging from < MDL (0.86 ng/g SOM) to 417570 ng/g SOM. The average concentration of Σ endosulfans in all background soil samples (n=57) was 23 ng/g SOM (SD±43 ng/g SOM), and ranged from < MDL (0.02 ng/g SOM) to 2620 ng/g SOM. The average concentration of PeCB for all sites studied (n=53) was 0.61 ng/g SOM (SD±0.51 ng/g SOM), ranging from < MDL (0.038 ng/g SOM) to 36 ng/g SOM.

For comparison, Table 1 additionally includes previously published data for some legacy POPs (PCBs, PBDEs and HCB) collected during the same sampling campaign (Schuster et al., 2011). Table A4 additionally includes concentration ranges for endosulfans, PeCB and SCCPs from other localities around the world. While a direct comparison is somewhat difficult because of differences in methodologies, it is encouraging to note that the concentration ranges reported in the literature are generally similar to the levels measured in this study (Table A4). The average concentrations of SCCPs in all soil samples were highest among all compounds listed in Table 1, followed by Σ_{31} PCBs (6±5 ng/g SOM). This pattern was consistent also for each subset of samples listed in Table 1 (UK, Norway, GL, and WL). Σ Endosulfans came out third for all soils combined (32 ± 43 ng/g SOM) including the Norwegian, UK and WL sub-sets. For all soils combined Concentration of, HCB (1 ± 1 ng/g SOM) was were approximately twice that ofmore or less at the same level as PeCB (0.61 ± 0.51 ng/g SOM) which in turn exceeded that of and Σ_5 PBDEs (0.6 ± 0.8 ng/g SOM) for all soils combined. The same pattern was evident for Norwegian together with and-WL and GL soils, but not for UK and GL soils. Here PeCB was found at the same level as exceeded both HCB and Σ_5 PBDEs in UK and GL soils. (Table 1).

209 Table 1:

3.2 Correlation analysis

211 **3.2.1** Removal of outliers

210

217

- 212 Outliers were removed prior to the correlation analysis. Results for PeCB (Birkenes, WL,
- Norway), Σ endosulfans (endosulfan sulfate, β -endosulfan) and Σ_{31} PCBs (Onsøy, WL, Norway),
- 214 SCCPs (Tamokdalen, GL, Norway) and Σ₅PBDEs (Okehampton, WL, UK) were consequently
- 215 excluded (A2.8.1). Correlation analysis was then performed on the full remaining dataset
- (Table 2), as well as individually for UK, Norway, GL, and WL soils (Tables A6-A9).

3.2.2 Correlations between compound groups

- For all soils combined, a highly significant correlation was found between the Σ endosulfans
- and HCB (r=0.82, p<0.001), Σ endosulfans and Σ_{31} PCBs (r=0.73, p<0.001) as well as between
- 220 Σendosulfans and PeCB (r=0.58, p<0.001) (Table 2). These results show that Σendosulfans
- share some similarities in their distribution and fate in these background soils with other POPs
- known to be relatively volatile (i.e. log K_{OA} < ~10, see Fig. 1a). The correlation between
- Σ_5 PBDEs (log K_{OA} >~10) and Σ endosulfans was significant, but weaker (r=0.38, p<0.01).
- 224 Significant correlations were also found between PeCB and HCB (r=0.63, p<0.001) and
- between PeCB and Σ_{31} PCBs (r=0.62, p<0.001) for all soils combined (Table 2), which is
- attributed to similarities in their predicted phase partitioning and distribution in soil (Fig. 1a).
- 227 PeCB and Σ_5 PBDEs displayed a weaker correlation for all soils (r=0.49, p<0.001) (Table 2),
- 228 which might be partly explained by a more divergent distribution partitioning behaviour as
- the Σ_5 PBDEs are less volatile (Fig. 1a, b).
- 230 SCCPs were not significantly correlated with any other POPs listed in Table 2 nor in any sub-
- sets (Tables A6-A9) with the notable exceptions of a weak and significant correlation with
- Σ_5 PBDEs for all soils combined (r=0.27, p<0.05) (Table 2) as well as for GL soils only (r=0.50,
- p<0.05) (Table A8). These results indicate that SCCPs are comparatively less prone to LRAT
- 234 (see also Fig. 2c, f).

Figure 1.

235236

237

238

239

240

241

242

243

244

245

246

247

248

249

250

251

252

253

254

255

256

257

258

259

260

261

262

263

264

than soil characteristics.

3.2.3 Correlation with site variables

It is well established that POPs and SOM are typically correlated in soils (e.g. Seth et al., 1999) which is also observed in this dataset (Table 2). For all GL soils (n=24), the SOM content varied from 11 to 93 % (w/w), while the SOM content ranged from 13 to 98 % within the WL soils (n=34). A confounding factor in the correlation analysis is the strong and significant correlation between SOM and BC for all soils combined (r=0.63, p<0.001) (Table 2). BC is a by-product of incomplete combustion of fossil fuels and biomass (Schmidt and Noack, 2000), and POPs may have the potential to be emitted with BC from an emission source (Nam et al., 2008a). SOM is composed of amorphous organic matter (AOM) and carbonaceous matter such as BC (Nam et al., 2008a). The sorption capacity of carbonaceous materials (BC) is considered to be remarkably higher compared to AOM, and BC is thought to be responsible for a large part of the sorption of organic contaminants in soils (Cornelissen et al., 2005; Lohmann et al., 2005). Consequently, correlations between POPs and SOM should be evaluated with caution as high correlations between POPs and SOM may in part be due to the sorption properties of BC. Another confounding variable is bulk density (BD) which is inversely proportional to SOM; hence SOM rich soil has low BD. Highest correlations between BC and the "new" POPs for all soils combined (Table 2) were seen with PeCB (r=0.57, p<0.001) followed by Σ endosulfans (r=0.46, p<0.001). Strong and significant correlation was also seen between SOM and Σendosulfans (r=0.80, p<0.001) albeit weaker for SOM and PeCB (r=0.70, p<0.001). In contrast to Σ endosulfans, PeCB is known to be emitted from a variety of industrial sources and combustion processes (Bailey et al., 2009; Doring et al., 1992). As PeCB experienced the strongest correlation with BC among all POPs listed in Table 2, this indicates that PeCB and BC may to some extent be co-emitted from various combustion processes. SCCPs showed no significant correlations (p>0.05) to any of the soil parameters, i.e. BD, BC and SOM, either for all soils combined nor for any sub-sets (Table 2, A6-A9). This indicates that

the distribution of SCCPs in soil is probably more influenced by proximity to sources, rather

Table 2:

265

266

267

268

269

270

272

273

274

275

276

277

278

279

280

281

282

283

284

285

286

287

288

289

290

291

292

293

294

295

ΣEndosulfans, endosulfan constituents and distribution within the soil subsets

Within the Σ endosulfans, endosulfan sulfate was the dominant component and was detected in all samples except one, while the α - and the β - endosulfan isomers were detected in 13 and 17 samples, respectively (Table 1, Table A5). Endosulfan sulfate contributed with 97 % to the average concentration of Σ endosulfans, while α -and β - endosulfan isomers contributed with

271 \sim 1.4 % and \sim 1.6 %, respectively.

> The elevated concentration of endosulfan sulfate, compared to α and β -endosulfan, is because the former is the effective transformation product of the parent endosulfan compounds in soils (Antonious et al., 1998; Walse et al., 2003). The α/β ratio of technical mixtures are reported to be 7/2 or 2/1, depending on the mixture (Weber et al., 2010). For sites where both α - and β -endosulfan were above MDL (n=7), the observed ratio in soils ranged from 0.1 to 0.9. These findings clearly imply a depletion of α -endosulfan in soils, relative to in the technical mixtures. We believe this depletion in part can be explained by Figure 1, which shows that α endosulfan is more volatile and less prone to atmospheric deposition (Fig. 1a) as well as more prone to evaporate from soils (Fig. 1b) in comparison to β-endosulfan and endosulfan sulfate, which are relatively more prone to erosion and leaching, respectively. These findings are consistent with a study by Rice and co-workers (2002) where α -endosulfan was found to be more volatile than β-endosulfan in soils. Furthermore, while strong and significant correlations are seen between Σendosulfans and endosulfan sulfate (r=0.99, p<0.001) as well as between Σ endosulfans and β -endosulfan (r=0.91, p<0.001) in background soils, there is no statistical significance between Σ endosulfans and α -endosulfan (r=0.49, p>0.05) which indicates divergent behaviour and fate of the latter isomer (Table 2). However, as α and β endosulfan were often below or close to MDL, the focus in the following is on Σ endosulfans.

> Results for the Σ endosulfan concerning the UK and Norwegian soil revealed essentially the same distribution pattern with respect to the soil parameters, i.e. BC, BD and SOM (Table A6 and A7). Several studies have revealed that the forest may act as filter for airborne contaminants by enhancing the rate of deposition (Horstmann and McLachlan, 1998; Wania and McLachlan, 2001). The average endosulfan concentration in WL soil was higher than in GL soil with 34 ng/SOM and 0.71 ng/g SOM, respectively (Table 1). Our data therefore suggest that Σ endosulfans are influenced by the forest filter effect, although the partitioning

properties (Fig. 1a) for β -endosulfan in particular are slightly outside the forest filter "window" (i.e. log K_{OA} and log K_{AW} around 9-10 and -2 to -3, respectively) as proposed by Wania and McLachlan (2001). A confounding factor, which might help to explain the enrichment of Σ endosulfans in WL soils compared to GL soils, is that the forest soils are higher in SOM (Table A1). The capacity of WL soils rich in SOM (average SOM 0.66 g/g dw) to retain Σ endosulfans may thus exceed that of GL soils (average SOM 0.44 g/g dw). The strong and significant correlation between Σ endosulfans and SOM is displayed in Figure 2g). Here the concentration of Σ endosulfans increased with amount of SOM both for GL and WL soil.

296

297

298

299

300

301

302

303

304

305

306

307

308

309

310

311

312

313

314

315

316

317

318

319

320

Figure 2a) and 2d) display the latitudinal distribution of Σ endosulfans expressed on SOM and on a dry weight basis, respectively. The concentration of Σ endosulfans for both WL and GL soil tends to increase towards 60 °N, followed by a slight decrease at higher latitudes, irrespective of the normalization. The elevated concentrations of Σ endosulfans seen at mid-latitudes (~60°N) may in part be caused by an orographic effect, which enhances wet deposition in southern parts of Norway compared to central and northern Norway (Aamot et al., 1996; Becker et al., 2011). The average concentration of Σ endosulfans for the Norwegian sites (n=32) was also higher than for the UK sites (n=24), with 3 and $\frac{2}{1}$ ng/g SOM, respectively (Table 1). An orographic effect on the spatial distribution of Σ endosulfan concentrations also seems plausible as Σ endosulfans are among the more water-soluble substances in our data set (Fig. 1a). Σ Endosulfans are furthermore positively correlated with latitude in GL soils, albeit with limited significance (r=0.47, p<0.05) (Table A8), yet negatively correlated (and not significant) for the Norwegian sub-set (Table A7). It is interesting to note that α -endosulfan is slightly positively correlated with latitude (r=0.24), whereas β-endosulfan as the more watersoluble substance is slightly negatively correlated (r=-0.11) for all soils combined (Table 2). However, none of these two correlations nor any correlations between average precipitation rates and Σ endosulfans, including its constituents, were significant (Tables 2 and A6-A9).

Figure 2 (a-i)

321

322

323

324

325

326

327

328

329

330

331

332

333

334

335

336

337

338

339

340

341

342

343

344

345

346

347

348

349

350

351

3.4 Pentachlorobenzene (PeCB) and distribution within soil subsets

The average concentration of PeCB in UK soils (n=21) was about twice somewhat higher than the average concentration for the Norwegian soils (n=31) with 0.82 ng/g SOM and 0.51 ng/g SOM, respectively (Table 1). The higher concentration of PeCB in UK soils may be due to proximity to past or ongoing source regions (Bailey et al., 2009; Nam et al., 2008a). It is interesting to note that this pattern is different from the one previously reported for HCB, where concentrations in Norwegian soils were found to exceed UK soils by a factor of about two (Table 1) in spite of close similarities in their partitioning behaviour (Fig. 1). Taken together with the strong association noted between PeCB and BC in soils, this provides further support for our hypothesis that PeCB may be co-emitted with BC from combustion processes. Combustion processes are also implicated as major emission sources of PeCB in the literature (Bailey et al., 2009; Tiernan et al., 1983). Studies have furthermore revealed that fly-ash from combustion processes contain PeCB (Liu et al., 2013; Nie et al., 2011). Furthermore, it is interesting to note that the correlation between PeCB and SOM was slightly stronger in Norwegian soils (r=0.80, p<0.001) (Table A7) compared to UK soils (r=0.71, p<0.001) (Table A6) which suggests that SOM may be somehow more important in controlling the occurrence in background soils in more remote regions of this transect. The average concentration of PeCB in GL soils (n=19) was more or less at the same level as WL soils (n=33), i.e. 0.64 ng/g SOM (Table 1). This suggests that PeCB is not affected by the forest filter effect. PeCB showed significant correlation with BC for GL soil (r=0.66, p<0.01) (Table A8), but to a lesser extent with BC in WL soil (r=0.41, p<0.05) (Table A9). This may be explained by the lower and weaker correlation between BC and SOM in WL soil (r=0.42, p<0.05) (Table A9) compared to in GL soil (r=0.79, p<0.001) (Table A8). The latter could be due to a higher influence of fresh input of organic matter containing organic carbon in forest soils, versus carbonaceous carbon (BC) (Agarwal and Bucheli, 2011), causing a possible dilution of BC in WL soils. Figure 2b) and 2e) present the latitudinal distribution of PeCB on a SOM and dry weight basis,

respectively. Concentrations of PeCB based on ng/g SOM for both WL and GL soil were

generally somewhat higher between ~ 50-55°N (>1.50.8 ng/g SOM in most samples),

compared to higher latitudes (typically less than 1.00.5 ng/g SOM) (Fig. 2b). Nonetheless, the

concentration of PeCB is more or less uniformly distributed with exception from of two one sites (Fig. 2b). Figure 2h) shows the concentration of PeCB expressed on dry weight basis versus the content of SOM in all soil samples. The figure clearly illustrates how the concentrations of PeCB for both WL and GL soil increase with increasing amounts of SOM. This increase furthermore appears particularly steep at low SOM content (~30%). These findings are supported by the strong and significant correlation between PeCB and SOM seen for all soils and each sub-set (Tables 2, A6-A9).

3.5 Short-chain chlorinated paraffins (SCCPs) and distribution within soil subsets

- The average concentration of SCCPs for the UK sites was approximately almost twice the average concentration for the Norwegian sites, with 50-31 ng/g SOM and 22-18 ng/g SOM, respectively (Table 1). The higher concentrations found in UK soil may reflect closer proximity to source regions (van der Gon et al., 2007). This is consistent with findings by Barber et al., (2005), who noted a significant temporal and spatial variability of SCCPs in UK air.
- No sign of a forest filter effect could be observed as the average SCCP concentration in GL soils was 59–38 ng/g SOM, which is more than close to 3 times higher than the average concentration in WL soils (1417 ng/g SOM) (Table 1).
 - Figures 2c) and 2f) illustrate the concentration of SCCPs as a function of latitude on an ng/g SOM and on a dry weight basis, respectively. The results show that the higher concentrations are typically seen at lower latitudes and that no sites at higher latitudes (> 62°N) experienced concentrations of SCCPs above MDL. This further suggests that SCCPs have a limited potential for LRAT as levels in soils decline relatively fast with increasing distance from suspected source regions. However, previous studies by Reth et al., (2006) and Tomy et al., (1999) show that SCCPs have been found in biota and sediments in the Arctic. Figure 2i) displays the concentrations of SCCPs on a dry weight basis versus SOM content (%) which further confirms the lack of correlation between SCCPs and SOM (Table 2, A6-A9).

4 Conclusion

The occurrence of Σendosulfans, PeCB, and SCCPs in background surface soils from Western
 Europe has been evaluated by the use of statistical analyses, distribution maps and mobility

plots in soil, as well as data on concentrations of legacy POPs (PCBs, HCB and PBDEs). Overall, the statistical analysis and mobility plots showed that Σ endosulfans and PeCB share many similarities with selected old POPs in their distribution in background soils. Moreover, PeCB and Σ endosulfans were highly correlated with each other, and soil characteristics as SOM and BC were both important in explaining their occurrence and distribution. Concentrations of Σ endosulfans peaked in areas experiencing elevated precipitation rates and were strongly dominated by endosulfan sulfate (~97%). PeCB was found with higher concentrations in UK soil compared to Norwegian soil. A particularly strong association between BC and PeCB suggests that emissions of PeCB in part may be attributed to various combustion processes, and therefore retained closer to source regions. SCCPs do not share many similarities with Σ endosulfans and PeCB nor most legacy POPs in these background soils, with the exception of PBDEs. A steep decline in SCCP concentrations with increasing latitude indicates a limited potential for long-range atmospheric transport. Further, the lack of correlation between soil concentrations and key soil characteristics, such as SOM, indicate that the occurrence of SCCPs is largely dictated by proximity to source regions. This study also demonstrates that simple chemical distribution and mobility plots (chemical space maps) are useful as complementary tools to assess whether any inferences made from statistical analysis of observations are reasonable from a mechanistic standpoint.

Appendix A. Supplementary data

400 Supplementary data available.

Acknowledgements

381

382

383

384

385

386

387

388

389

390

391

392

393

394

395

396

397

398

399

401

402

403

404

405

406

407

This study was financed by the Research Council of Norway (183437 and 196191). The authors thank Sabine Eckhardt for valuable assistance on the ECMWF data, provided by the Norwegian MetOffice. We would also like to thank our colleagues at NILU (A. Borgen, H. Gundersen, H. Lunder, K. Davanger, I.S. Krogseth and M. Ghebremeskel) for support, together with Andy Sweetman and Rosalinda Gioia for their valuable contributions.

5 References

- 409 Aamot, E., Steinnes, E., and Schmid, R. (1996). Polycyclic aromatic hydrocarbons in Norwegian forest soils: Impact of long range atmospheric transport. *Environmental Pollution* **92**, 275-280.
- 411 Agarwal, T., and Bucheli, T. D. (2011). Adaptation, validation and application of the chemo-thermal oxidation method to quantify black carbon in soils. *Environmental Pollution* **159**, 532-538.
- 413 Antonious, G. F., Byers, M. E., and Snyder, J. C. (1998). Residues and fate of endosulfan on field-grown 414 pepper and tomato. *Pesticide Science* **54**, 61-67.
- Bailey, R. E., van Wijk, D., and Thomas, P. C. (2009). Sources and prevalence of pentachlorobenzene in the environment. *Chemosphere* **75**, 555-564.
 - Barber, J. L., Sweetman, A. J., Thomas, G. O., Braekevelt, E., Stern, G. A., and Jones, K. C. (2005). Spatial and temporal variability in air concentrations of short-chain (C-10-C-13) and medium-chain (C-14-C-17) chlorinated n-alkanes measured in the UK atmosphere. *Environmental Science & Technology* **39**, 4407-4415.
- Becker, L., Scheringer, M., Schenker, U., and Hungerbuhler, K. (2011). Assessment of the environmental persistence and long-range transport of endosulfan. *Environmental Pollution* **159**, 1737-1743.
 - Borghini, F., Grimalt, J. O., Sanchez-Hernandez, J. C., and Bargagli, R. (2005). Organochlorine pollutants in soils and mosses from Victoria Land (Antarctica). *Chemosphere* **58**, 271-278.
 - Cornelissen, G., Gustafsson, O., Bucheli, T. D., Jonker, M. T. O., Koelmans, A. A., and Van Noort, P. C. M. (2005). Extensive sorption of organic compounds to black carbon, coal, and kerogen in sediments and soils: Mechanisms and consequences for distribution, bioaccumulation, and biodegradation. *Environmental Science & Technology* **39**, 6881-6895.
 - Dalla Valle, M., Jurado, E., Dachs, J., Sweetman, A. J., and Jones, K. C. (2005). The maximum reservoir capacity of soils for persistent organic pollutants: implications for global cycling. *Environmental Pollution* **134**, 153-164.
 - Doring, J., Damberg, M., Gamradt, A., and Oehme, M. (1992). Screening method based on the determination of perchlorinated aromatics for surface soil contaminated by copper slag containing high-levels of polychlorinated dibenzofurans and dibenzo-p-dioxines. *Chemosphere* **25**, 755-762.
 - European Chemicals Bureau (2008). "European Union Risk Assessment Report: Alkanes, C10-13, chloro, updated version." European Chemicals Bureau, Luxemburg.
 - Gawor, A., and Wania, F. (2013). Using quantitative structural property relationships, chemical fate models, and the chemical partitioning space to investigate the potential for long range transport and bioaccumulation of complex halogenated chemical mixtures. *Environmental Science-Processes & Impacts* **15**, 1671-1684.
 - Gluge, J., Bogdal, C., Scheringer, M., Buser, A. M., and Hungerbuhler, K. (2013). Calculation of Physicochemical Properties for Short- and Medium-Chain Chlorinated Paraffins. *Journal of Physical and Chemical Reference Data* **42**.
 - Gouin, T., Mackay, D., Webster, E., and Wania, F. (2000). Screening chemicals for persistence in the environment. *Environmental Science & Technology* **34**, 881-884.
 - Hassanin, A., Breivik, K., Meijer, S. N., Steinnes, E., Thomas, G. O., and Jones, K. C. (2004). PBDEs in European background soils: Levels and factors controlling their distribution. *Environmental Science & Technology* **38**, 738-745.
- Hassanin, A., Lee, R. G. M., Steinnes, E., and Jones, K. C. (2005). PCDD/Fs in Norwegian and UK soils: Implications for sources and environmental cycling. *Environmental Science & Technology* **39**, 4784-4792.
- Horstmann, M., and McLachlan, M. S. (1998). Atmospheric deposition of semivolatile organic compounds to two forest canopies. *Atmospheric Environment* **32**, 1799-1809.
- Houde, M., Muir, D. C. G., Tomy, G. T., Whittle, D. M., Teixeira, C., and Moore, S. (2008).
 Bioaccumulation and trophic magnification of short- and medium-chain chlorinated paraffins

- in food webs from Lake Ontario and Lake Michigan. *Environmental Science & Technology* **42**, 3893-3899.
- Krogseth, I. S., Breivik, K., Arnot, J. A., Wania, F., Borgen, A. R., and Schlabach, M. (2013). Evaluating the environmental fate of short-chain chlorinated paraffins (SCCPs) in the Nordic environment using a dynamic multimedia model. *Environmental Science-Processes & Impacts* 15, 2240-2251.
- Li, N. Q., Wania, F., Lei, Y. D., and Daly, G. L. (2003). A comprehensive and critical compilation, evaluation, and selection of physical-chemical property data for selected polychlorinated biphenyls. *Journal of Physical and Chemical Reference Data* **32**, 1545-1590.

- Liu, G. R., Liu, W. B., Cai, Z. W., and Zheng, M. H. (2013). Concentrations, profiles, and emission factors of unintentionally produced persistent organic pollutants in fly ash from coking processes. *Journal of Hazardous Materials* **261**, 421-426.
- Lohmann, R., MacFarlane, J. K., and Gschwend, P. M. (2005). Importance of black carbon to sorption of native PAHs, PCBs, and PCDDs in Boston and New York, Harbor sediments. *Environmental Science & Technology* **39**, 141-148.
- Meijer, S. N., Ockenden, W. A., Sweetman, A., Breivik, K., Grimalt, J. O., and Jones, K. C. (2003). Global distribution and budget of PCBs and HCB in background surface soils: Implications or sources and environmental processes. *Environmental Science & Technology* **37**, 667-672.
- Meijer, S. N., Steinnes, E., Ockenden, W. A., and Jones, K. C. (2002). Influence of environmental variables on the spatial distribution of PCBs in Norwegian and UK soils: Implications for global cycling. *Environmental Science & Technology* **36**, 2146-2153.
- Moeckel, C., Nizzetto, L., Di Guardo, A., Steinnes, E., Freppaz, M., Filippa, G., Camporini, P., Benner, J., and Jones, K. C. (2008). Persistent Organic Pollutants in Boreal and Montane Soil Profiles: Distribution, Evidence of Processes and Implications for Global Cycling. *Environmental Science & Technology* **42**, 8374-8380.
- Muir, D., Stern, G., and Tomy, G. (2000). Chlorinated paraffins. *In* "The Handbook of Environmental Chemistry" (J. Paasivirta, ed.), Vol. 3. Springer-Verlag, Berlin, Heidelberg.
 - Nam, J. J., Gustafsson, O., Kurt-Karakus, P., Breivik, K., Steinnes, E., and Jones, K. C. (2008a). Relationships between organic matter, black carbon and persistent organic pollutants in European background soils: Implications for sources and environmental fate. *Environmental Pollution* **156**, 809-817.
- Nam, J. J., Thomas, G. O., Jaward, F. M., Steinnes, E., Gustafsson, O., and Jones, K. C. (2008b). PAHs in background soils from Western Europe: Influence of atmospheric deposition and soil organic matter. *Chemosphere* **70**, 1596-1602.
- Nie, Z. Q., Zheng, M. H., Liu, W. B., Zhang, B., Liu, G. R., Su, G. J., Lv, P., and Xiao, K. (2011). Estimation and characterization of PCDD/Fs, dl-PCBs, PCNs, HxCBz and PeCBz emissions from magnesium metallurgy facilities in China. *Chemosphere* **85**, 1707-1712.
 - Reth, M., Ciric, A., Christensen, G. N., Heimstad, E. S., and Oehme, M. (2006). Short- and medium-chain chlorinated paraffins in biota from the European Arctic differences in homologue group patterns. *Science of the Total Environment* **367**, 252-260.
 - Rice, C. P., Nochetto, C. B., and Zara, P. (2002). Volatilization of trifluralin, atrazine, metolachlor, chlorpyrifos, alpha-endosulfan, and beta-endosulfan from freshly tilled soil. *Journal of Agricultural and Food Chemistry* **50**, 4009-4017.
 - Roots, O., Roose, A., Kull, A., Holoubek, I., Cupr, P., and Klanova, J. (2010). Distribution pattern of PCBs, HCB and PeCB using passive air and soil sampling in Estonia. *Environmental Science and Pollution Research* 17, 740-749.
- 503 Schmidt, M. W. I., and Noack, A. G. (2000). Black carbon in soils and sediments: Analysis, distribution, 504 implications, and current challenges. *Global Biogeochemical Cycles* **14**, 777-793.
- Schuster, J. K., Gioia, R., Moeckel, C., Agarwal, T., Bucheli, T. D., Breivik, K., Steinnes, E., and Jones, K. C. (2011). Has the Burden and Distribution of PCBs and PBDEs Changed in European Background Soils between 1998 and 2008? Implications for Sources and Processes. *Environmental Science & Technology* **45**, 7291-7297.

- 509 Seth, R., Mackay, D., and Muncke, J. (1999). Estimating the organic carbon partition coefficient and its variability for hydrophobic chemicals. *Environmental Science & Technology* **33**, 2390-2394.
- 511 Shen, L., Wania, F., Lei, Y. D., Teixeira, C., Muir, D. C. G., and Bidleman, T. F. (2005). Atmospheric 512 distribution and long-range transport behavior of organochlorine pesticides in north America. 513 *Environmental Science & Technology* **39**, 409-420.
- Sverko, E., Tomy, G. T., Marvin, C. H., and Muir, D. C. G. (2012). Improving the Quality of Environmental
 Measurements on Short Chain Chlorinated Paraffins to Support Global Regulatory Efforts.
 Environmental Science & Technology 46, 4697-4698.
- Tiernan, T. O., Taylor, M. L., Garrett, J. H., Vanness, G. F., Solch, J. G., Deis, D. A., and Wagel, D. J. (1983).
 Chlorodibenzodioxines, chlorodibenzofurans and related-compounds in the effluents from combustion processes. *Chemosphere* 12, 595-606.
- Tomy, G. T., Stern, G. A., Lockhart, W. L., and Muir, D. C. G. (1999). Occurrence of C-10-C-13 polychlorinated n-alkanes in Canadian midlatitude and arctic lake sediments. *Environmental Science & Technology* **33**, 2858-2863.
 - Tomy, G. T., Stern, G. A., Muir, D. C. G., Fisk, A. T., Cymbalisty, C. D., and Westmore, J. B. (1997). Quantifying C-10-C-13 polychloroalkanes in environmental samples by high-resolution gas chromatography electron capture negative ion high resolution mass spectrometry. *Analytical Chemistry* **69**, 2762-2771.
- 527 U.S. EPA (2011). Estimation Programs Interface Suite(TM) for Microsoft Windows, v 4.1., Vol. 2013.
- 528 UNECE (2010). "The 1998 Protocol on Persistent Organic Pollutants, including the Amendments 529 Adopted by the Parties on 18 December 2009."
 - UNECE ad hoc expert group on POPs (2003). "Short chain chlorinated paraffins (SCCPs) Substance Dossier. Final Draft II."
- 532 UNEP (2009). "Stockholm Convention on Persistent Organic Pollutants (POPs) as amended in 2009.
 533 Text and annexes."
- UNEP (2011). "Adoption of an amendment to Annex A," Geneva.

523

524

525

526

530

531

535

536

537

538

539

540

541

542

543

544

545

546

547

548

549

- UNEP (2012). "Report of the Persistent Organic Pollutants Review Committee on the work of its eighth meeting". Stockholm Convention on Persistent Organic Pollutants.
- van der Gon, H. D., van het Bolscher, M., Visschedijk, A., and Zandveld, P. (2007). Emissions of persistent organic pollutants and eight candidate POPs from UNECE-Europe in 2000, 2010 and 2020 and the emission reduction resulting from the implementation of the UNECE POP protocol. *Atmospheric Environment* **41**, 9245-9261.
- Walse, S. S., Scott, G. I., and Ferry, J. L. (2003). Stereoselective degradation of aqueous endosulfan in modular estuarine mesocosms: formation of endosulfan gamma-hydroxycarboxylate. *Journal of Environmental Monitoring* **5**, 373-379.
- Wang, X. T., Zhang, Y., Miao, Y., Ma, L. L., Li, Y. C., Chang, Y. Y., and Wu, M. H. (2013). Short-chain chlorinated paraffins (SCCPs) in surface soil from a background area in China: occurrence, distribution, and congener profiles. *Environmental Science and Pollution Research* **20**, 4742-4749.
- Wania, F., and Dugani, C. B. (2003). Assessing the long-range transport potential of polybrominated diphenyl ethers: A comparison of four multimedia models. *Environmental Toxicology and Chemistry* **22**, 1252-1261.
- Wania, F., and McLachlan, M. S. (2001). Estimating the influence of forests on the overall fate of semivolatile organic compounds using a multimedia fate model. *Environmental Science & Technology* **35**, 582-590.
- Weber, J., Halsall, C. J., Muir, D., Teixeira, C., Small, J., Solomon, K., Hermanson, M., Hung, H., and Bidleman, T. (2010). Endosulfan, a global pesticide: A review of its fate in the environment and occurrence in the Arctic. *Science of the Total Environment* **408**, 2966-2984.
- Wong, F., Robson, M., Diamond, M. L., Harrad, S., and Truong, J. (2009). Concentrations and chiral signatures of POPs in soils and sediments: A comparative urban versus rural study in Canada and UK. *Chemosphere* **74**, 404-411.

560	Wong, F., and Wania, F. (2011). Visualising the equilibrium distribution and mobility of organic
561	contaminants in soil using the chemical partitioning space. Journal of Environmental
562	Monitoring 13 , 1569-1578.
563	Zhao, Z., Zeng, H., Wu, J., and Zhang, L. (2013). Organochlorine pesticide (OCP) residues in mountain
564	soils from Tajikistan. Environmental Science-Processes & Impacts 15, 608-616.

566	Figure captions in the order of appearance						
567	Figure 1: Chemical partitioning space maps for selected POPs, a) equilibrium phase distribution, b)						
568	mobility potential in soils (modified after Wong and Wania (2011). Data for the compound groups is						
569	referred to in section 2.4.						
570							
571	Figure 2 (a-i): Latitudinal distribution for Σ endosulfans, PeCB and SCCPs on ng/g SOM (a, b, c) and dry						
572	weight basis (d, e, f). Also included is the log concentration (ng g-1 dw) versus SOM (g, h, i). Woodland						
573	(WL, ▲) and grassland (GL, O) soil samples are plotted separately. One outlier for each compound was						
574	omitted (2.8.1).						

Figure 1 a), b)

1.5 fitting image

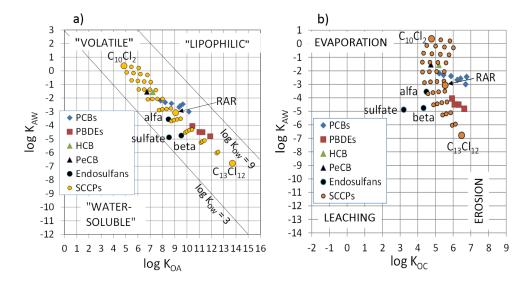


Table 1: Average soil concentrations \pm standard deviation and ranges, for Σ endosulfans and its constituents, PeCB and SCCPs for all soils combined (ng/g SOM). Samples which failed to meet the specified QA/QC criteria were excluded (see 3.1). The table also includes Σ_{31} PCBs, Σ_5 PBDEs and HCB (Schuster et al., 2011).

Aritmetric average ± standard deviation

Range (min-max)

	>MDL	Σendosulfans	endosulfan sulfate	α-endosulfan	β-endosulfan	PeCB	SCCPs		Σ_{31} PCBs	$\Sigma_5 PBDEs$	НСВ
All 2008	All 2008	2±4	2±4	0.02±0.02	0.04±0.1	0.6±0.5	24±72	All 2008	6±5	0.6±0.8	1±1
n = 57/53/58	n = 56/52/23	<0.022)-26	<0.003 ²⁾ -25	<0.0052)-0.08	<0.003 ²⁾ -0.7	<0.03 ²⁾ -3	<0.82)-417	n = 70	0.2-27	0.01-4	0.04-7
Norway	Norway	3±5	3±5	0.02±0.02	0.06±0.1	0.5±0.4	18±73	Norway	8±6	0.3±0.4	1±1
n= 32/32/32	n= 32/31/9	0.08-26	0.04-25	<0.0052)-0.07	<0.003 ²⁾ -0.7	<0.03 ²⁾ -2	<0.82)-417	n= 40	0.2-27	0.01-2	0.2-5
UK	UK	1±2	1±2	0.02±0.02	0.02±0.03	0.8±0.6	31±71	UK	5±3	1±1	0.8±1
n= 25/21/26	n= 24/21/14	<0.02 ²⁾ -10	<0.003 ²⁾ -10	<0.0052)-0.08	<0.003 ²⁾ -0.1	0.2-3	<0.92)-317	n= 30	0.5-10	0.1-4	0.04-7
GL	GL	0.7±0.9	0.6±0.9	0.02±0.02	0.01±0.01	0.6±0.7	38±103	GL	5±5	0.5±0.8	1±1
n= 23/20/24	n= 22/19/9	<0.02 ²⁾ -4	<0.003 ²⁾ -4	<0.0052)-0.08	<0.003²)-0.06	<0.03 ²⁾ -3	<0.92)-417	n= 30	0.2-23	0.02-4	0.06-7
WL	WL	3±5	3±5	0.02±0.02	0.06±0.1	0.6±0.4	14±37	WL	8±6	0.7±0.7	1±1
n= 34/33/34	n= 34/33/14	0.1-26	0.1-25	<0.005 ²⁾ -0.07	<0.003 ²⁾ -0.7	0.2-2	<0.82)-205	n= 40	0.6-27	0.01-3	0.04-5

¹⁾ Number (n) of sites for which Σ endosulfans, PeCB and SCCPs were determined, respectively.

²⁾ One or more samples were below MDL (½ MDL used for statistical summaries).

³⁾ Data from (Schuster et al., 2011)

Figure 2 (a-i): Latitudinal distribution for Σ endosulfans, PeCB and SCCPs on ng/g SOM (a, b, c) and dry weight basis (d, e, f). Also included is the log concentration (ng g⁻¹ dw) versus SOM (g, h, i). Woodland (WL, \blacktriangle) and grassland (GL, O) soil samples are plotted separately. One outlier for each compound was omitted (2.8.1).

